

## A NEW ROUTE TO OPTICALLY PURE *cis*- AND *trans*-2,5-DISUBSTITUTED PYRROLIDINES

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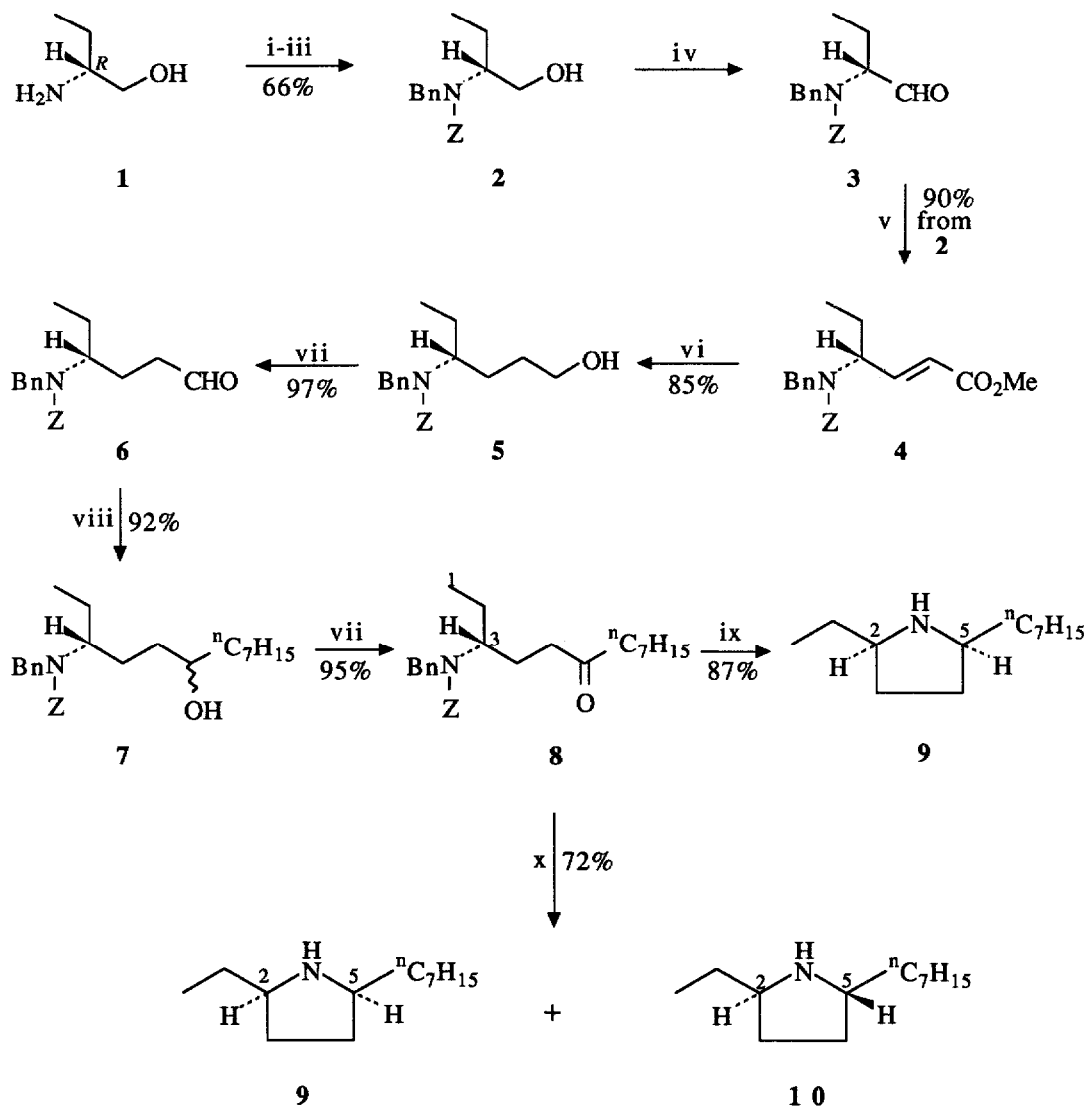
**Abstract** : A new route to optically pure *cis*- and *trans*-2,5-disubstituted pyrrolidines is described and is illustrated by the enantiospecific synthesis of (2*R*, 5*S*)-(+)-2-ethyl-5-heptylpyrrolidine **9** and (2*R*, 5*R*)-(-)-2-ethyl-5-heptylpyrrolidine **10** starting from (*R*)-(-)-2-amino-1-butanol **1**.

Several 2,5-disubstituted pyrrolidines, pyrrolines and 2-alkyl-6-methylpiperidines are known to occur among the predominant constituents of the venoms of ants belonging to different species of the genera *Solenopsis* and *Monomorium*. These compounds exhibit a wide range of physiological activities which include necrotic, hemolytic, phytotoxic, insecticidal, antibacterial, and antifungal properties.<sup>1</sup> A number of syntheses, mostly in the racemic mode,<sup>2</sup> have been reported for these bases. The only enantioselective syntheses of representative ant pyrrolidine toxins are those reported by Husson *et al.*<sup>3</sup> and by Shiosaki and Rapoport.<sup>4</sup>

In connection with our work relating to chiral synthesis of alkaloids using amino acids as the source of chirality,<sup>5</sup> we describe here a new route to optically pure *cis*- and *trans*-2,5-disubstituted pyrrolidines. Our approach is illustrated by the synthesis of *cis*-(2*R*, 5*S*)-2-ethyl-5-heptylpyrrolidine **9** and *trans*-(2*R*, 5*R*)-2-ethyl-5-heptylpyrrolidine **10** starting from commercially available (*R*)-(-)-2-amino-1-butanol **1** (see Scheme).

The amino group of the starting (*R*)-(-)-2-amino-1-butanol **1** was fully protected through successive introduction of a benzyl (Bn) and a benzyloxycarbonyl (Z) group to give the alcohol **2**. The aldehyde **3**, obtained by Swern oxidation<sup>6</sup> of **2**, was subjected to Wittig olefination with methyl(triphenylphosphoranylidene)acetate yielding the  $\alpha,\beta$ -unsaturated ester **4**. Reduction of the conjugated ester group of **4** with sodium borohydride in the presence of lithium chloride<sup>7</sup> provided the saturated alcohol **5** which was oxidised with pyridinium chlorochromate (PCC) to the aldehyde **6**<sup>8</sup>. Grignard alkylation of **6** with *n*-C<sub>7</sub>H<sub>15</sub>MgBr led to the mixture of alcohols **7** which, on PCC oxidation, afforded the ketone **8**. The latter was converted to the 2,5-disubstituted pyrrolidines **9** and **10** by way of two different hydrogenation procedures. Thus, hydrogenation of **8** with 10% Pd-C in methanol removed the amino protecting groups with concomitant ring closure and reduction to give exclusively the *cis*-2,5-disubstituted pyrrolidine **9**, namely, (2*R*, 5*S*)-(+)-2-ethyl-5-heptylpyrrolidine.

## Scheme



**Reagents and reaction conditions :** i. PhCHO, MgSO<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, molecular sieves 4Å, 25°C; ii. NaBH<sub>4</sub>, MeOH, at 0°C and then 2h at room temperature; iii. PhCH<sub>2</sub>OCOC<sub>1</sub>, NaOH, H<sub>2</sub>O-dioxan (1:1), 5°C; iv. Me<sub>2</sub>SO, (COCl)<sub>2</sub>, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, -78°C; v. Ph<sub>3</sub>P=CHCO<sub>2</sub>Me, THF, 25°C; vi. NaBH<sub>4</sub>-LiCl (1:1), EtOH-THF (4:3), 25°C; vii. PCC, CH<sub>2</sub>Cl<sub>2</sub>, molecular sieves, 4Å, 25°C; viii. n-C<sub>7</sub>H<sub>15</sub>MgBr, Et<sub>2</sub>O, 25°C; ix. H<sub>2</sub>, 10% Pd-C, MeOH, 25°C; x. HCOONH<sub>4</sub>, 10% Pd-C, MeOH, reflux, 1min.

By contrast, when the reductive deprotection was carried out with ammonium formate<sup>9</sup> in the presence of 10% Pd-C in refluxing methanol under nitrogen, a mixture of *cis*- and *trans*-2,5-disubstituted pyrrolidines **9** and **10** was obtained. The two isomers **9** and **10**, in the ratio 3:2, could be separated by silica gel column chromatography and their structure was established by the <sup>1</sup>H nmr studies of their N-benzyl derivatives using the method of Chan and Hill.<sup>10</sup>

The synthetic (+)-**9** and (-)-**10** had spectral data<sup>11</sup> which fully matched those previously reported.<sup>3</sup> Their enantiomeric homogeneity was established through HPLC and 400 MHz <sup>1</sup>H n.m.r. analysis of their (S)-(-)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)phenylacetyl derivatives (Mosher amides).<sup>12</sup>

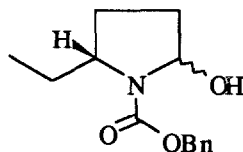
In conclusion, the route we have described here seems to be sufficiently general in as much as it should be amenable to an array of amino acids leading to the synthesis of a variety of differently substituted 2,5-dialkylpyrrolidines. Furthermore, it should be possible to extend this method to 2,6-disubstituted piperidines and our current efforts are directed toward this goal.

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#### References and notes :

1. G.A. Adrouny, V.J. Derbes and R.C. Jung, *Science*, (1959), **130**, 449; D.C. Buffkin and F.F. Russel, *Toxicon*, (1972), **10**, 526; A.B. Attygalle and E.D. Morgan, *Chem. Soc. Rev.*, (1984), **13**, 245; J.L. Clement, M. Lemaire, C. Lange, G. Lhommet, J.P. Celerier, J.J. Basselier and P. Casier, *Chem. Abstarct*, (1986), **104**, 202330j.
2. G. Massiot and C. Delaude, "The Alkaloids", Ed. A. Brossi, Academic press, N.Y. (1986), **27**, pp 269-322; A. Numata and T. Ibuka, *ibid.* (1987), vol. **31**, pp 193-315 and references cited; J.W. Daly, T.F. Spande, N. Whittaker, R.J. Highet, D. Feigl, N. Nishimori, T. Tokuyama and C.W. Myers, *J. Nat. Prod.*, (1986), **49**, 265; V. Baliah, R. Jeyaraman and L. Chandrasekaran, *Chem. Rev.*, (1983), **83**, 379; W. Gessner, K. Takahashi, A. Brossi, M. Kowalski and M.A. Kaliner, *Hel. Chim. Acta.*, (1987), **70**, 2003; D. Bacos, J.J. Basselier, J.P. Celerier, C. Lange, E. Marx, G. Lhommet, P. Escoubas, M. Lemaire and J.L. Clement, *Tetrahedron Lett.*, (1988), **29**, 3061.
3. P.Q. Huang, S. Arseniyadis and H.P. Husson, *Tetrahedron Lett.*, (1987), **28**, 547; *ibid.*, (1988), **29**, 631; S. Arseniyadis, P.Q. Huang, D. Piveteau and H.P. Husson, *Tetrahedron*, (1988), **24**, 2457.
4. K. Shiosaki and H. Rapoport, *J. Org. Chem.*, (1985), **50**, 1229. See also J.L. Marco, *J. Heterocycl. Chem.*, (1986), **23**, 287 and 1059.
5. S. Jegham and B.C. Das, *Tetrahedron Lett.*, (1988), **35**, 4419; S. Jegham, J.L. Fourrey and B.C. Das, *Tetrahedron Lett.*, (accepted for publication).
6. A.J. Mancuso, S.L. Huang and D. Swern, *J. Org. Chem.*, (1978), **43**, 2480; K. Omura and D. Swern, *Tetrahedron*, (1978), **34**, 1651.

7. Y. Hamada, M. Shibata, T. Sugiura, S. Kato and T. Shioiri, *J. Org. Chem.*, (1987), **52**, 1252.
8. It is to be noted that, when the reaction sequence described for **1** to **6** was carried out with a singly protected amino group such as with a benzyloxycarbonyl (Z), the aldehyde corresponding to **6** (see Scheme) was obtained only in the cyclic form **6a** - an observation which is not without precedent in the literature [F.G. Salituro, N. Agarwal, T. Hoffmann and D.H. Rich, *J. Med. Chem.*, (1987), **30**, 286].

**6 a**

9. S. Ram and L. D. Spicer, *Tetrahedron Lett.*, (1987), **28**, 515; *Synth. Commun.*, (1987), **17**, 415.
10. R.K. Hill and T.H. Chan, *Tetrahedron*, (1965), **21**, 2015.
11. All reported yields are for purified materials isolated from column chromatography and all compounds gave spectroscopic data in agreement with the assigned structures. Selected physical data include the following.
- 8** : colourless oil;  $^1\text{H NMR}$  (80 MHz,  $\text{CDCl}_3$ )  $\delta$  0.80 (t, 6H, 2 x Me), 1.43 (m, 14H, 7 x  $\text{CH}_2$ ), 2.10 (m, 4H, 2 x  $\text{CH}_2$ ), 3.83 (m, 1H, H-3), 4.30 (s, 2H,  $\text{CH}_2\text{Ph}$ ), 5.13 (s, 2H,  $\text{CO}_2\text{CH}_2\text{Ph}$ ), 7.23 (m, 10H, 2 x Ph); EIMS  $m/z$  437( $\text{M}^+$ );  $[\alpha]_{\text{D}} -4^\circ$  (c 2,  $\text{CHCl}_3$ ).
- 9** : colourless oil;  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  : 0.86 (t, 3H, Me,  $J$  5.5), 1.02 (t, 3H, Me,  $J$  7.5), 1.14-1.48 (m, 10H), 1.68-1.94 (m, 4H), 1.94-2.30 (m, 4H), 3.43 (m, 2H, H-2, H-5).  $^{13}\text{C NMR}$  (50,2 MHz,  $\text{CDCl}_3$ )  $\delta$  : 11.7, 14.1 (2 x Me), 22.7, 27.6, 28.9, 29.4, 29.9, 30.5, 30.9, 32.1, 36.6 (9 x  $\text{CH}_2$ ), 59.5 (C-2), 61.0 (C-5). CIMS  $m/z$  198 ( $\text{MH}^+$ );  $[\alpha]_{\text{D}} +4^\circ$  (c 2,  $\text{CHCl}_3$ ), lit.<sup>3</sup>  $+7^\circ$  (c 0.5,  $\text{CHCl}_3$ ).
- 10** : colourless oil;  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  : 0.96 (t, 3H, Me,  $J$  6), 1.00 (t, 3H, Me,  $J$  7.5), 1.15-1.50(m, 10H), 1.50-2.00 (m, 4H), 2.20-2.90 (m, 4H), 3.41 (m, 2H, H-2, H-5).  $^{13}\text{C NMR}$  (50,2 MHz,  $\text{CDCl}_3$ )  $\delta$  : 11.4, 14.1 (2 x Me), 22.7, 27.3, 28.7, 29.3, 29.7, 31.8, 31.9, 32.2, 36.4 (9 x  $\text{CH}_2$ ), 58.2(C-2), 59.8 (C-5);  $[\alpha]_{\text{D}} -4^\circ$  (c 0.6,  $\text{CHCl}_3$ ), lit.<sup>3</sup>  $+4^\circ$  (c 2,  $\text{CHCl}_3$ ) for (+)-enantiomer.
12. J. A. Dale and H.S. Mosher, *J. Am. Chem. Soc.*, (1973), **95**, 512.

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